

# Towards Better Adhesive Design – Using Scientific Tools

## DMA, the Luth-Burgers Model & DoE

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**7<sup>th</sup> Afera  
Technical Seminar**



Brussels, 13-15 April

**2015**

# AZC - Our global map



- Production sites
- Sales, trading and representative offices
- Executive offices
- Science and Technology centers

# Largest Producer of Pine Chemicals



- **Largest Producer of Pine Chemicals**
  - **Hold 50% global capacity of crude tall oil production at 800,000 tons/year**
- **World class manufacturing facilities and practices**
  - **Favorable carbon footprint balance**
    - **Products range from 50-99% Bio-renewable content**
  - **75% fewer CO2 emissions than oil-based substitutes**
- **Committed to continuous improvement**
  - **ISO 14001**
  - **ISO 9000**
  - **OHSAS 18001**
  - **Responsible Care Program**



# Raw materials for Adhesives



Case & Carton Sealing



Non-Wovens



PSA Tape & Labels



Building & Construction



Transportation



Bookbinding



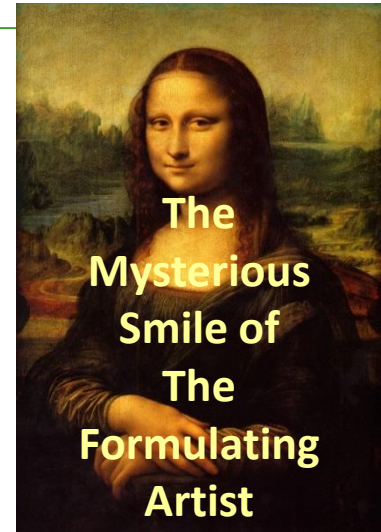
Product Assembly



Wood Assembly

# Key Challenges in Adhesive Design

- **Current adhesive design is an art**
  - Formulations may become *overly complex*
    - Too many “critical” raw materials
  - Development is *time-consuming*
    - With too often a dead-end road taken
  - Depending on the *experience* of a few
    - Not sustainable
  - Inability to adapt (quickly) to new technologies when they appear





# Adhesives – improving by using a model



## Step 1

- **Creating a model of PSA action**
  - shows the action(s) of the components
- **Components *interact***

## Step 2

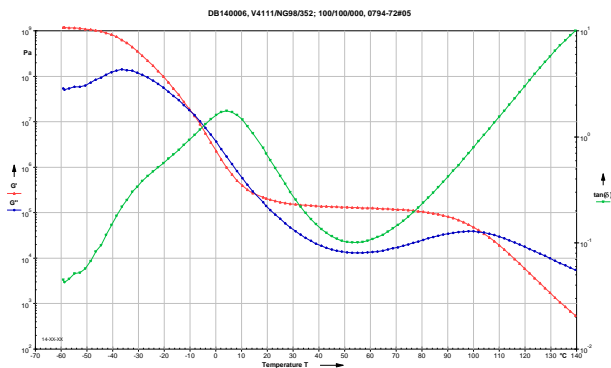
- **Map the *non-linear* interactions: DoE**

## Step 3

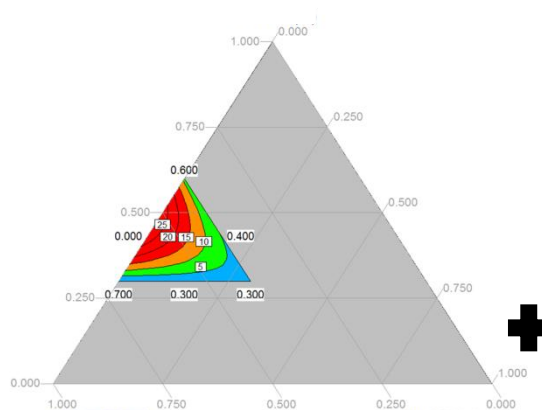
- **Create model for formulation**

# Adhesive Model Building – in pictures

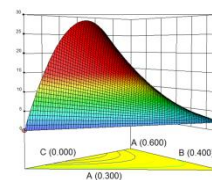
## DMA, HSP



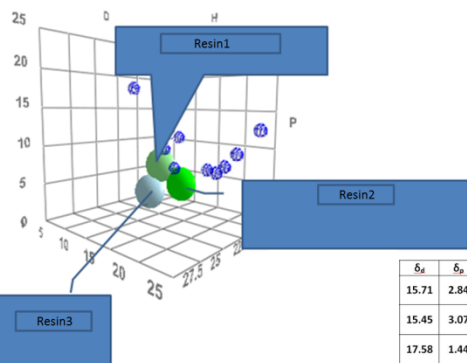
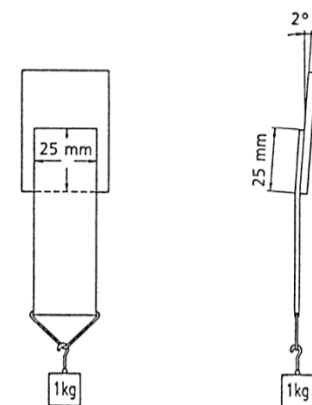
## DoE



180o Peel 20min SS (N)



## Application Data



$\delta_d$	$\delta_p$	$\delta_h$	Radius	Resin Type
15.71	2.84	7.89	8.5	Rosin ester
15.45	3.07	8.78	9.5	Terpene phenol
17.58	1.44	3.36	6.5	Polyterpene

# What is a PSA?

- **PSA's**
  - are materials that are aggressively and permanently tacky at room temperature and that firmly adhere to a variety of dissimilar surfaces upon mere contact without the need of more than finger or hand pressure
- **Hypothesis:**

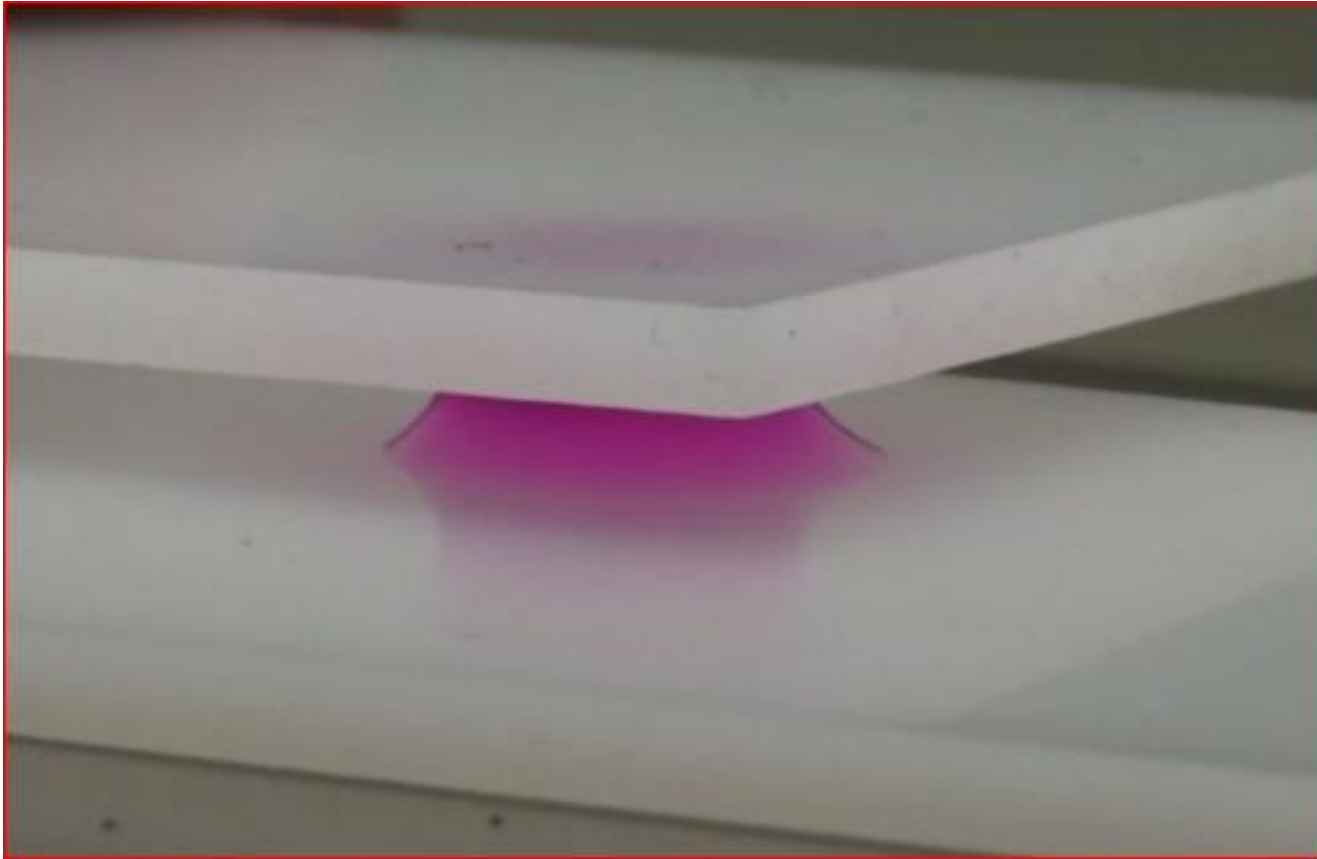
***Adhesion is less of a surface phenomenon, much more a bulk material property***



# Look what happens:

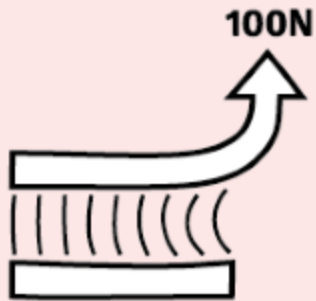


# Reconsidering old truths...

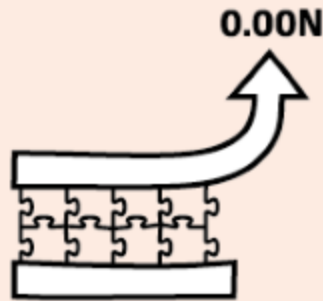


Water will *spontaneously* wet PE .....

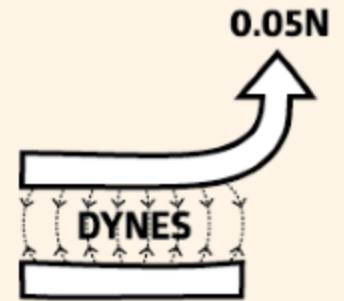
# Adhering - in pictures



Why does  
stuff stick?

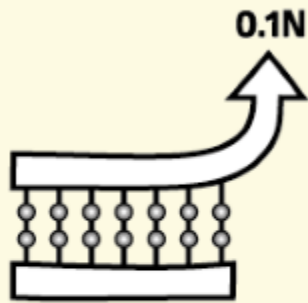


It's not  
mechanical

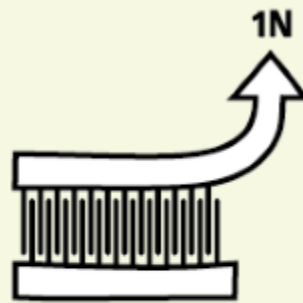


Surface energy's  
too weak

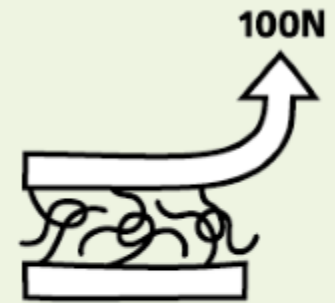
# Adhering - in pictures



It's not  
chemical bonds

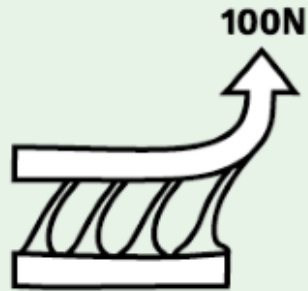


Intermingling  
helps

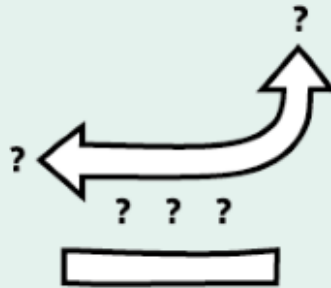


Entanglement  
is strong

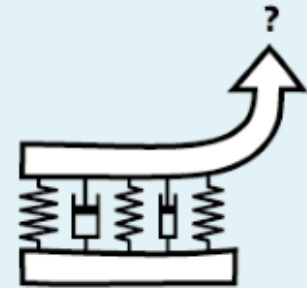
# Adhering - in pictures



Dissipation  
is strong



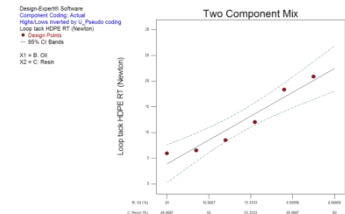
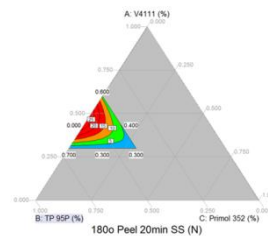
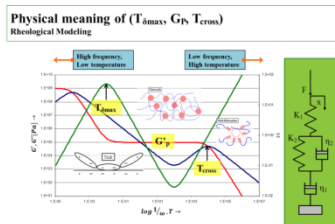
Measurement  
is tricky



Adhesion is  
a property of  
the system

# Modelling of Pressure Sensitive Adhesives using **DMA**

- Old adage “adhesion is a surface phenomenon” is not valid
  - Need to study **bulk** properties as well
- DMA viscoelastic profiles:
  - Study, explain & *predict* PSA behaviour
- *Ultimately: facilitate more intelligent formulation*

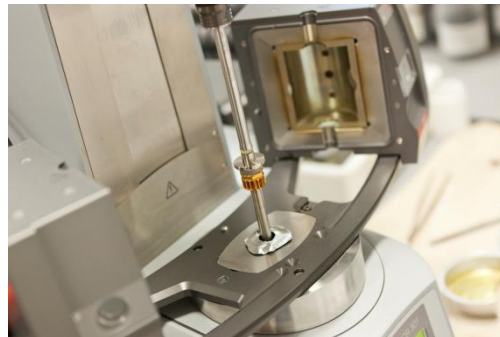
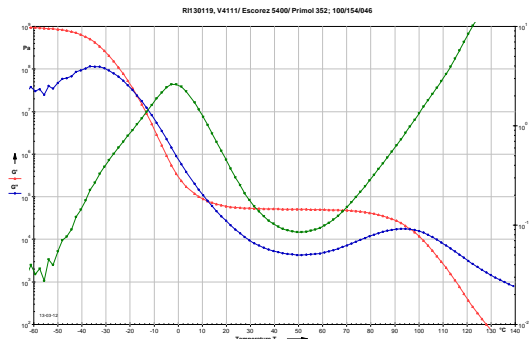




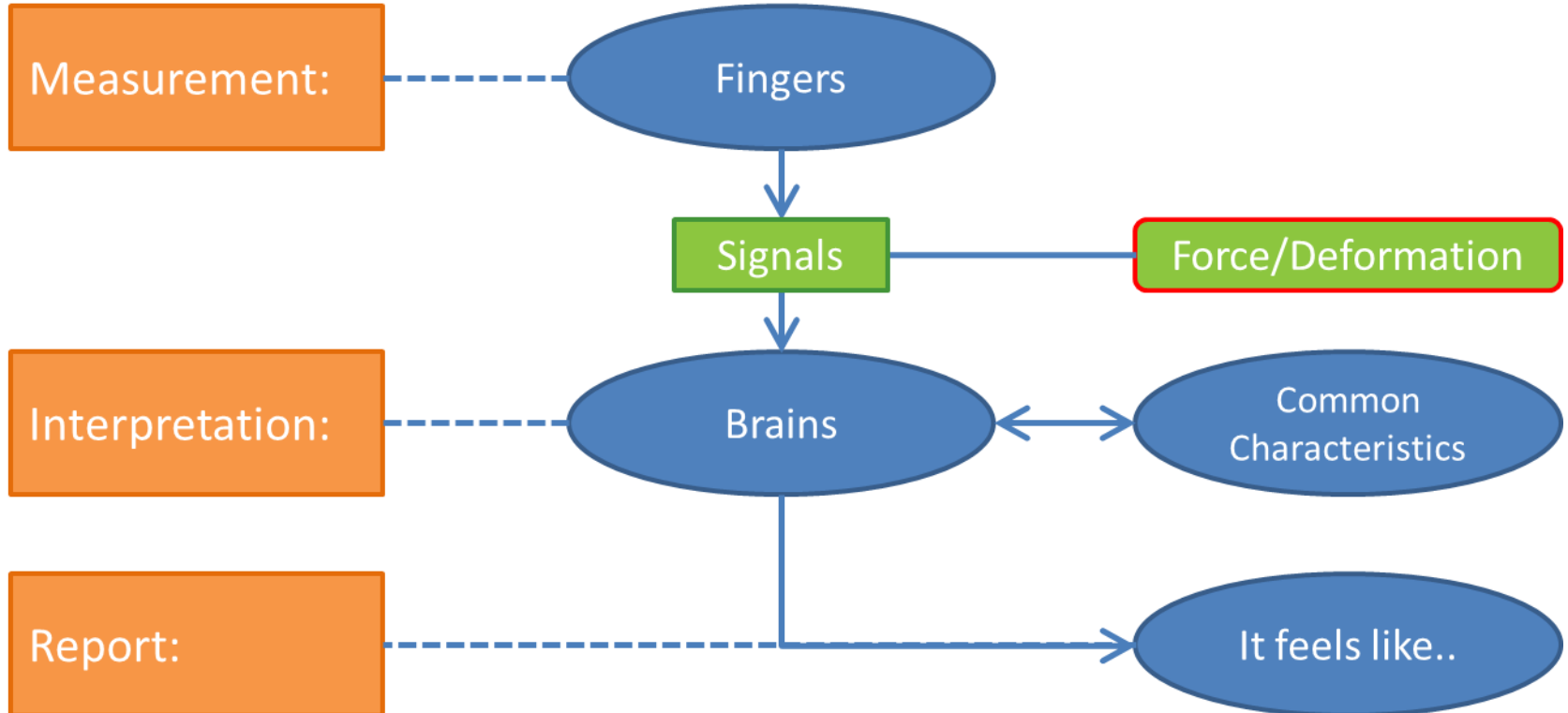
# The principle and result

## Dynamic Mechanical Analysis

- What is **D**ynamic **M**echanical **A**nalysis?
  - For a start it is:
- *“An Accurate Way of Feeling”*



### The feeling process



# The principle and result

## Dynamic Mechanical Analysis

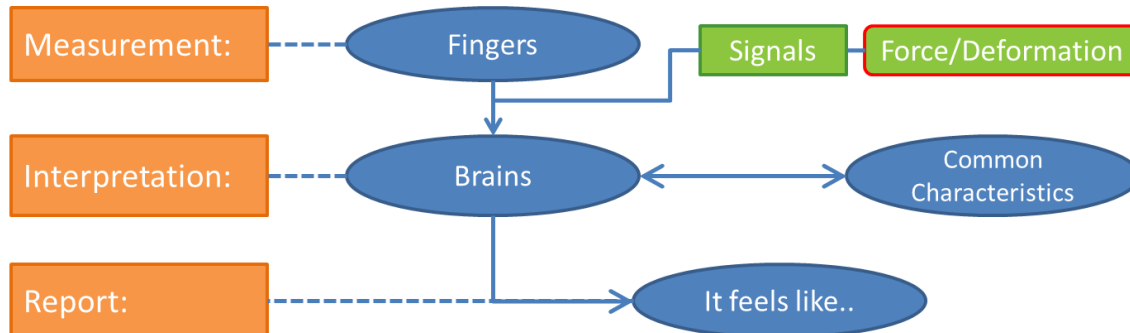
### Rheology (definition)

Rheos, (ῥέος) =flow of matter

The study of the relation between **forces** active in a material and the thereby induced **deformations**

(includes flow as well).

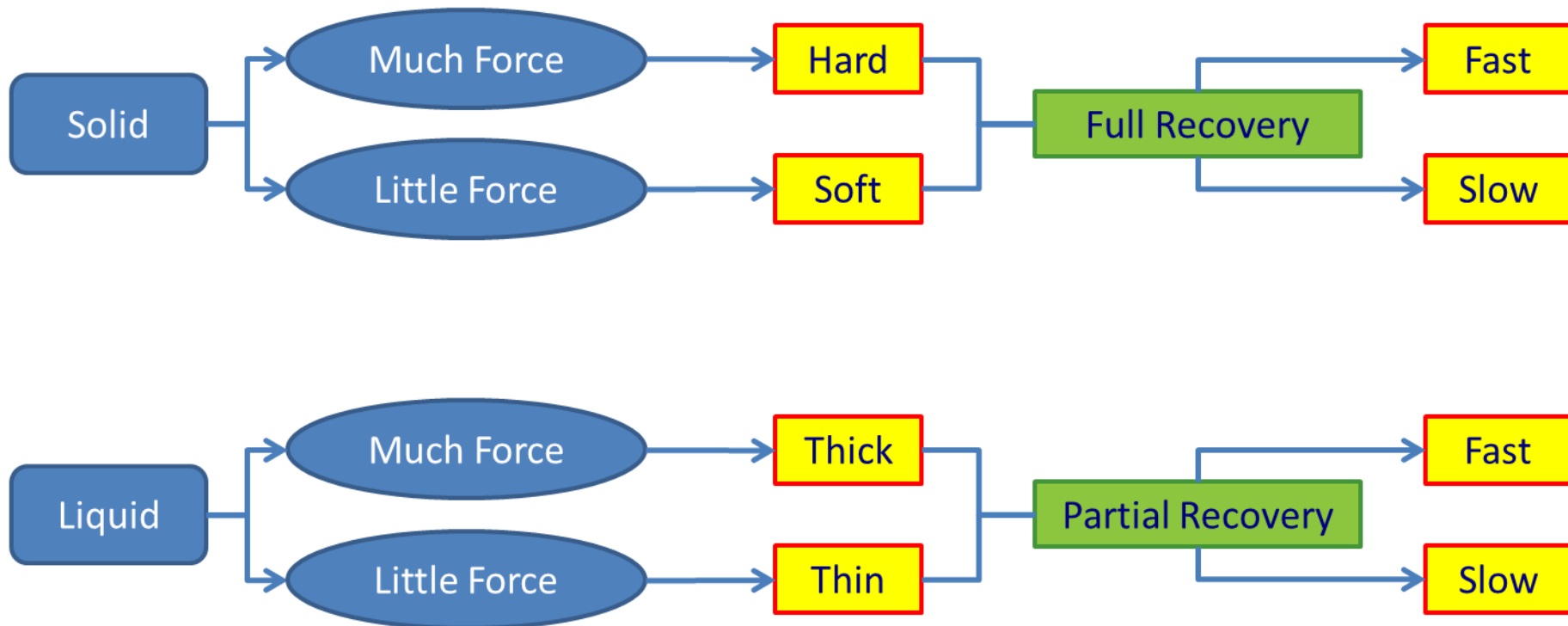
*“Feeling is doing Rheology”*



# The principle and result

## Dynamic Mechanical Analysis

### Distinctions by feeling



# DMA – how does the machine *feel* ?

Stationary geometry

PSA Sample

2mm thick sample  
8mm wide sample

Oscillating geometry

Reference Sample, ARES G2 Rheometer

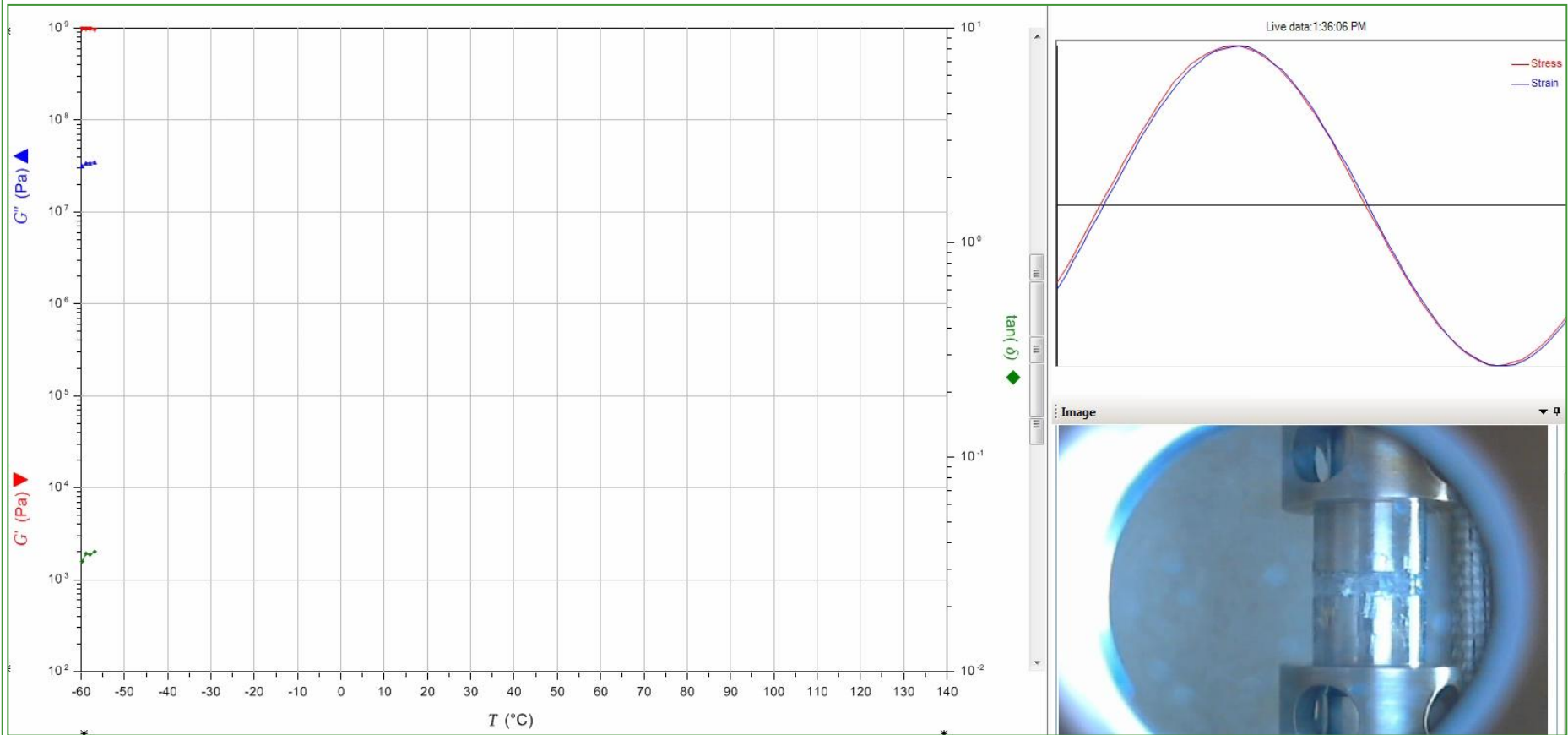
Copyright: D. Beekman, Arizona Chemical



1 CM



# DMA – Measurement (50x real time)

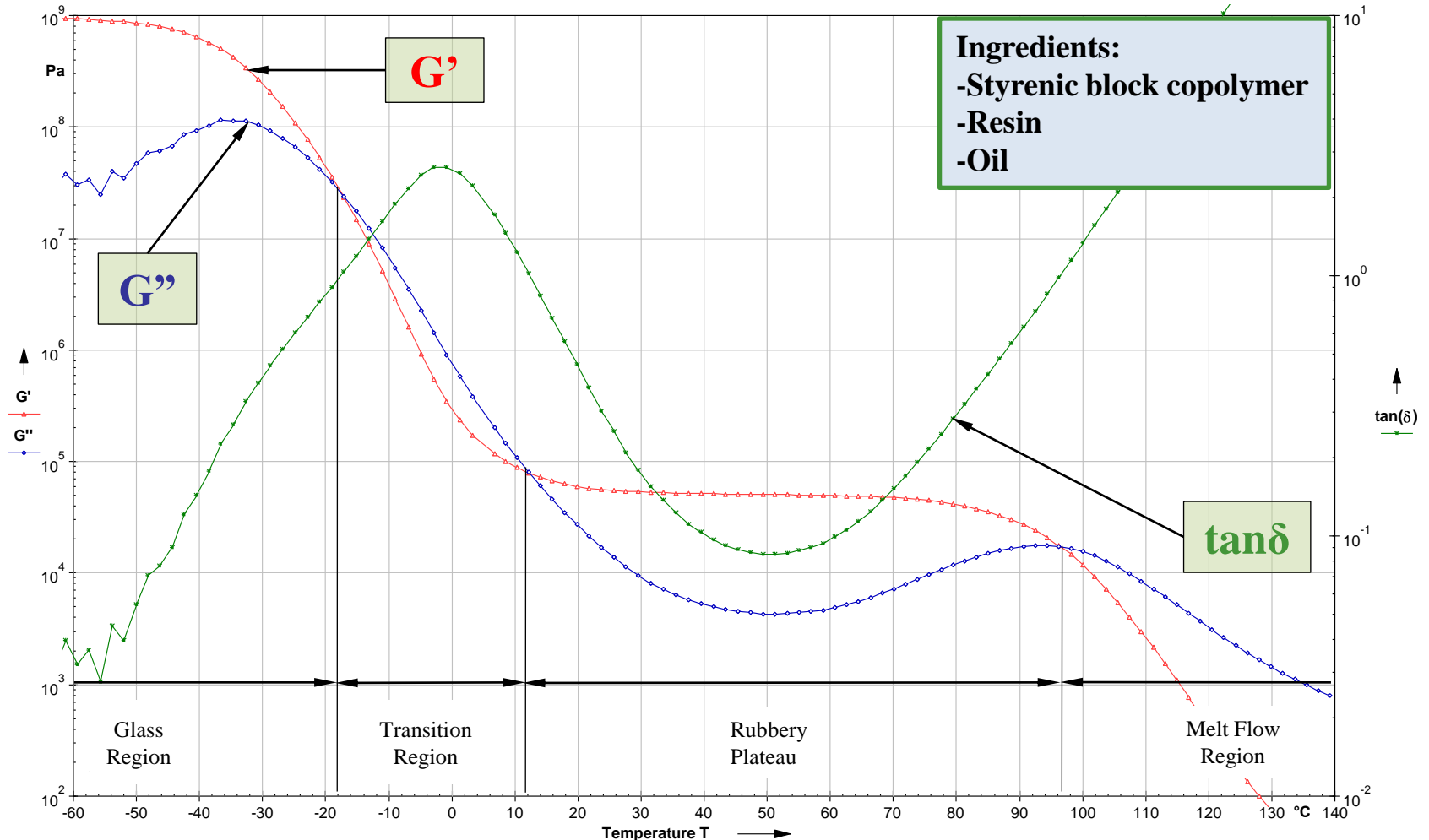




# Connection to Reality

## Dynamic Mechanical Analysis

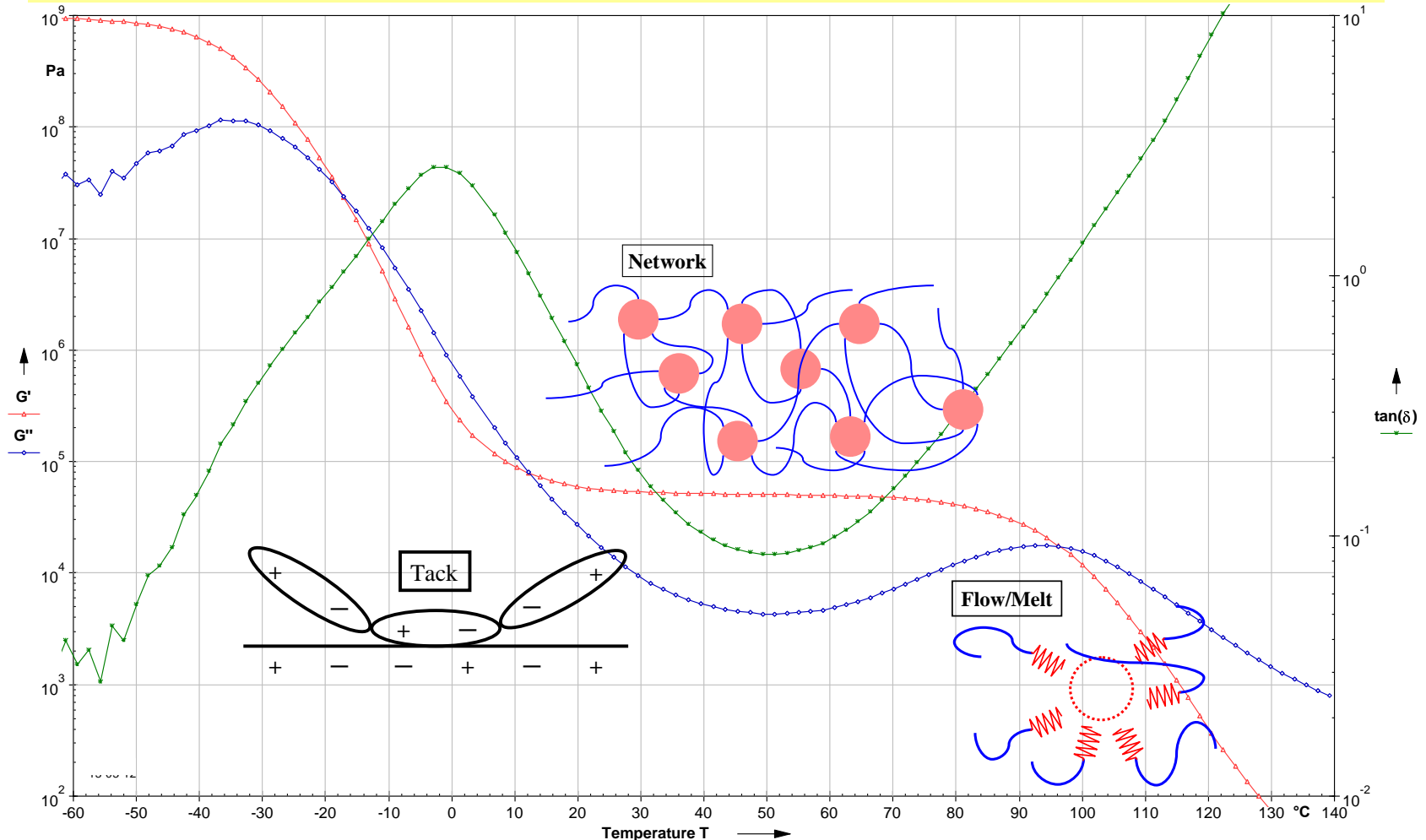
### Typical viscoelastic response of a “real Hot Melt Pressure Sensitive Adhesive system”



# Connection to Reality

## Dynamic Mechanical Analysis

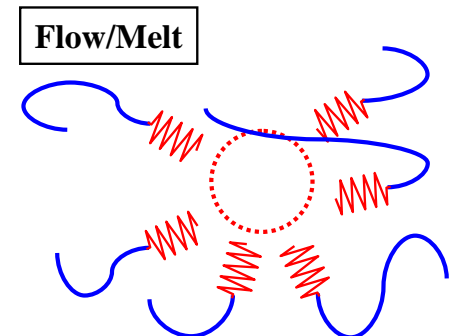
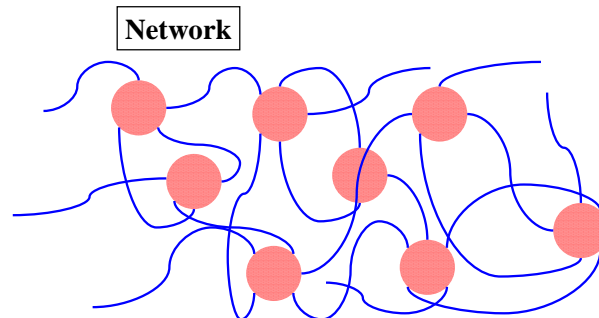
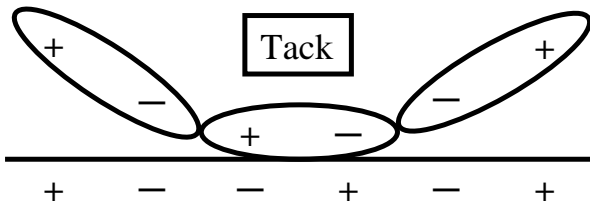
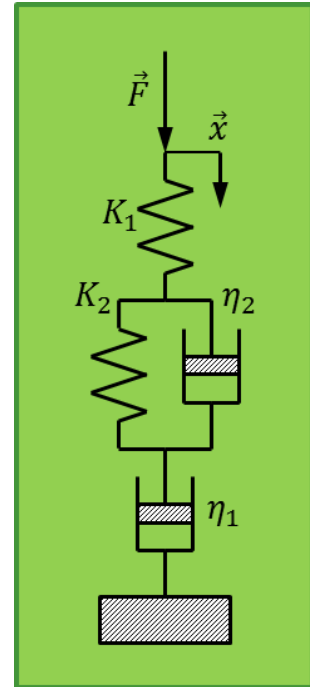
### Typical areas related to HMPSA characteristics



# Ideal (HM)PSA / Key Parameters

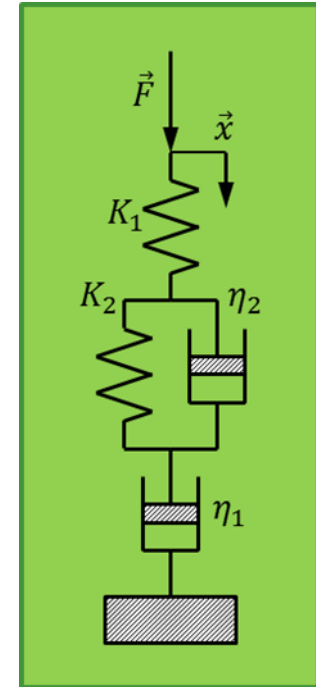
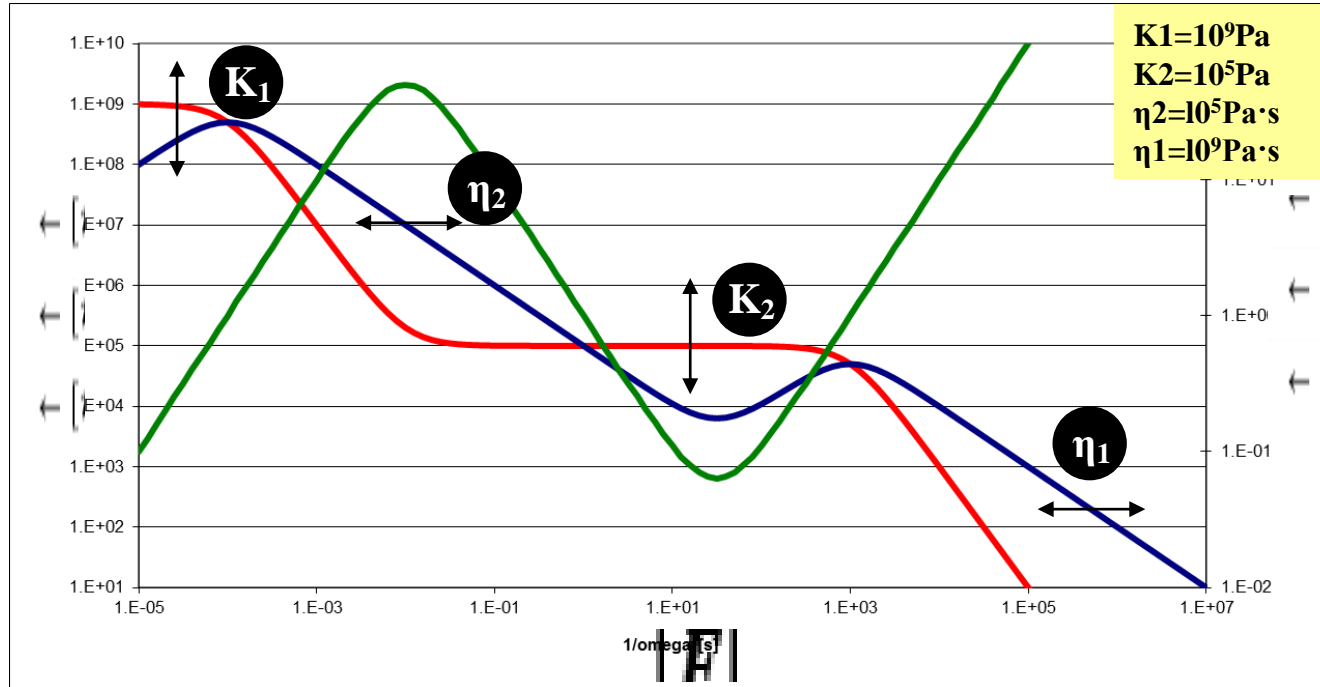
## Dynamic Mechanical Analysis

- **An SIS/Resin/Oil blend.**
  - Hard Styrenic domains connected by elastic isoprenic midblocks;  $K_2$
  - Local resin has temperature dependent damping effect on movement  $K_2$ ;  $\eta_2//K_2$
  - Free flow at high temperature;  $\eta_1$
  - Glassy at very low temp.;  $K_1$
- **Model is known as the “Burgers model”**



# Ideal HMPSA / Key Parameters

## The Luth-Burgers PSA Model



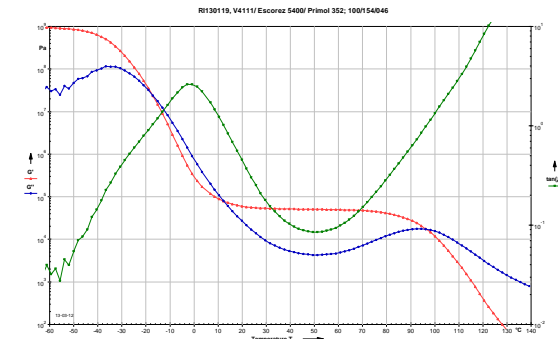
$$\frac{F}{x} = \frac{1}{\left(\frac{1}{K_1} + \frac{1}{K_2 + j\omega\eta_2} + \frac{1}{j\omega\eta_1}\right)} = \frac{-\omega^2\eta_2\eta_1K_1 + j\omega\eta_1K_1K_2}{(K_1K_2 - \omega^2\eta_1\eta_2) + j\omega(\eta_1(K_1 + K_2) + \eta_2K_1)}$$

$$\frac{A + jB}{C + jD} = \left(\frac{AC + BD}{C^2 + D^2}\right) + j\left(\frac{-AD + BC}{C^2 + D^2}\right) = G' + jG''$$

### Facts:

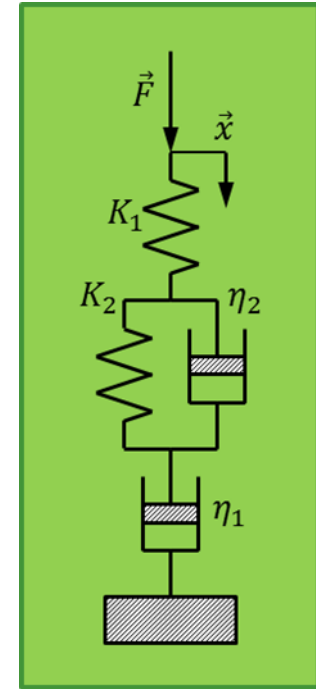
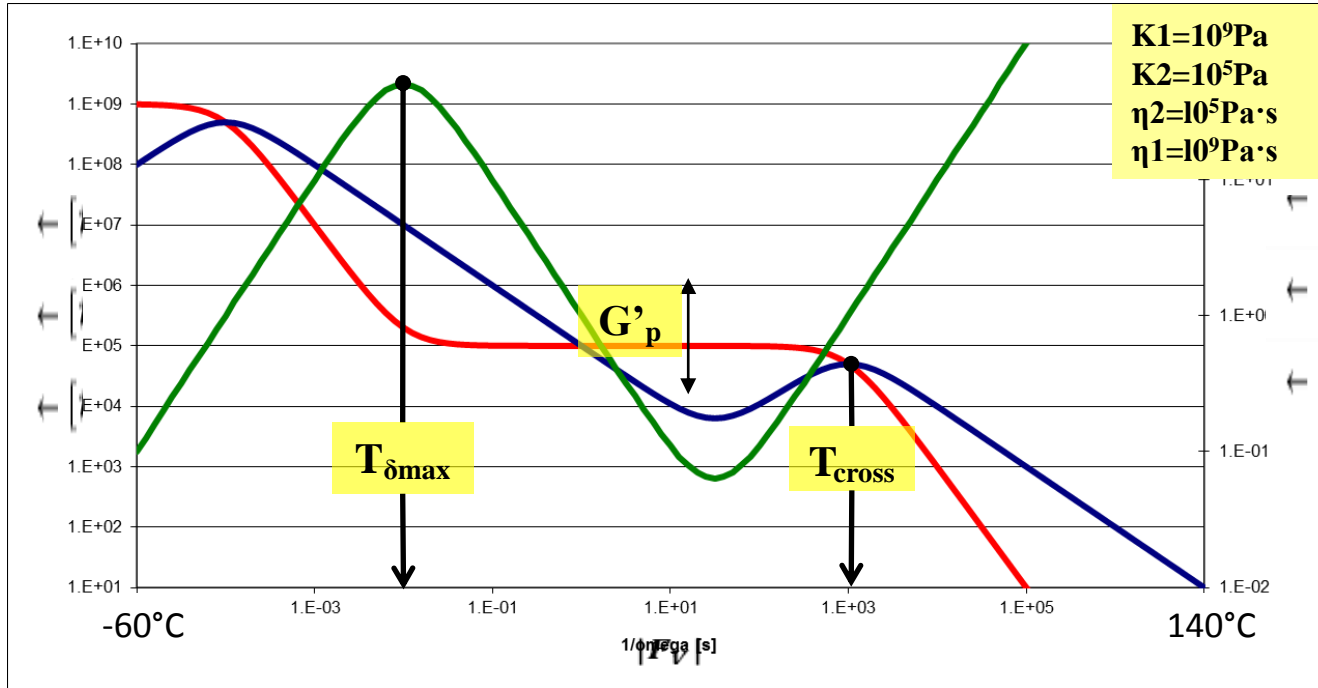
Fixed values for  $K_1$ ,  $K_2$ ,  $\eta_1$ ,  $\eta_2$ , represent an unique viscoelastic profile.

$G'(\omega)$  and  $G''(\omega)$  @  $T=\text{const.}$ ;  
 Introducing temperature dependent dashpots will result in  $G'(T)$  and  $G''(T)$



# Ideal HMPSA / Key Parameters

## The Luth-Burgers PSA Model

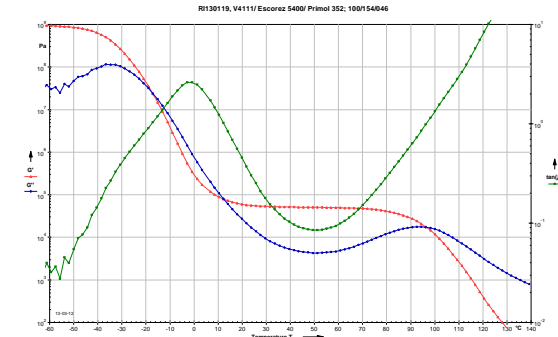


$$\frac{F}{x} = 1 / \left( \frac{1}{K_1} + \frac{1}{K_2 + j\omega\eta_2} + \frac{1}{j\omega\eta_1} \right) = \frac{-\omega^2\eta_2\eta_1K_1 + j\omega\eta_1K_1K_2}{(K_1K_2 - \omega^2\eta_1\eta_2) + j\omega(\eta_1(K_1 + K_2) + \eta_2K_1)}$$

$$\frac{A + jB}{C + jD} = \left( \frac{AC + BD}{C^2 + D^2} \right) + j \left( \frac{-AD + BC}{C^2 + D^2} \right) = G' + jG''$$

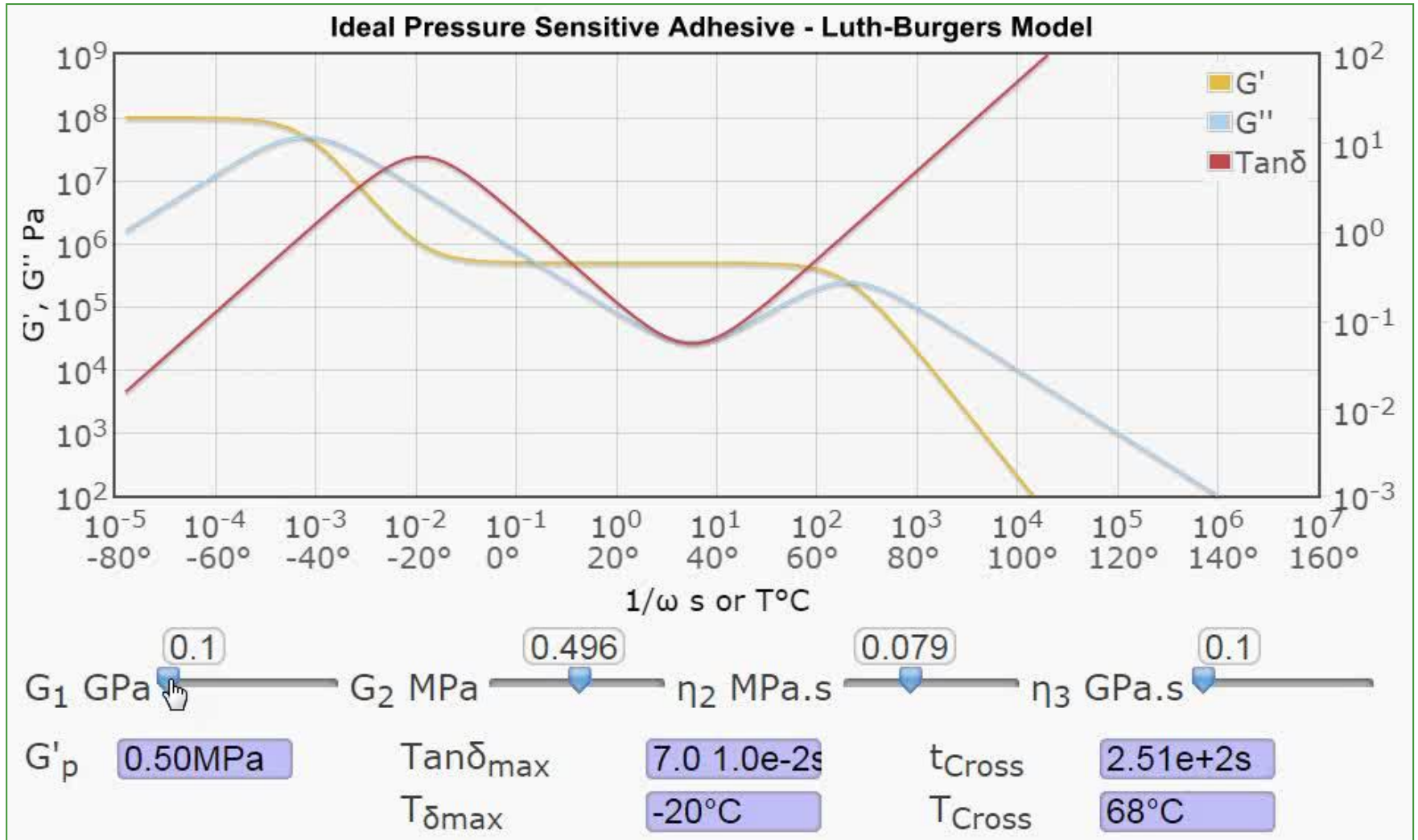
### Facts:

$\{(-), T_{\delta\max}, G'_p, T_{\text{cross}}\}; \{(K_1), K_2, \eta_1, \eta_2\}$   
 Temp. sweep;  $\eta(T) = \eta_0 \exp\{-A(T - T_{\text{ref}})\}$ .  
 i.e varying “log1/ω” or “T” will result, apart from a scaling factor, in a similar viscoelastic profile {WLF TTS}.



# Interactive version of the

# Luth-Burgers PSA model (Now available on the internet!)



Pictures and animations used with permission - Prof. Steven Abbott (<http://www.stevenabbott.co.uk/PracticalAdhesion/IdealPSA.html>)



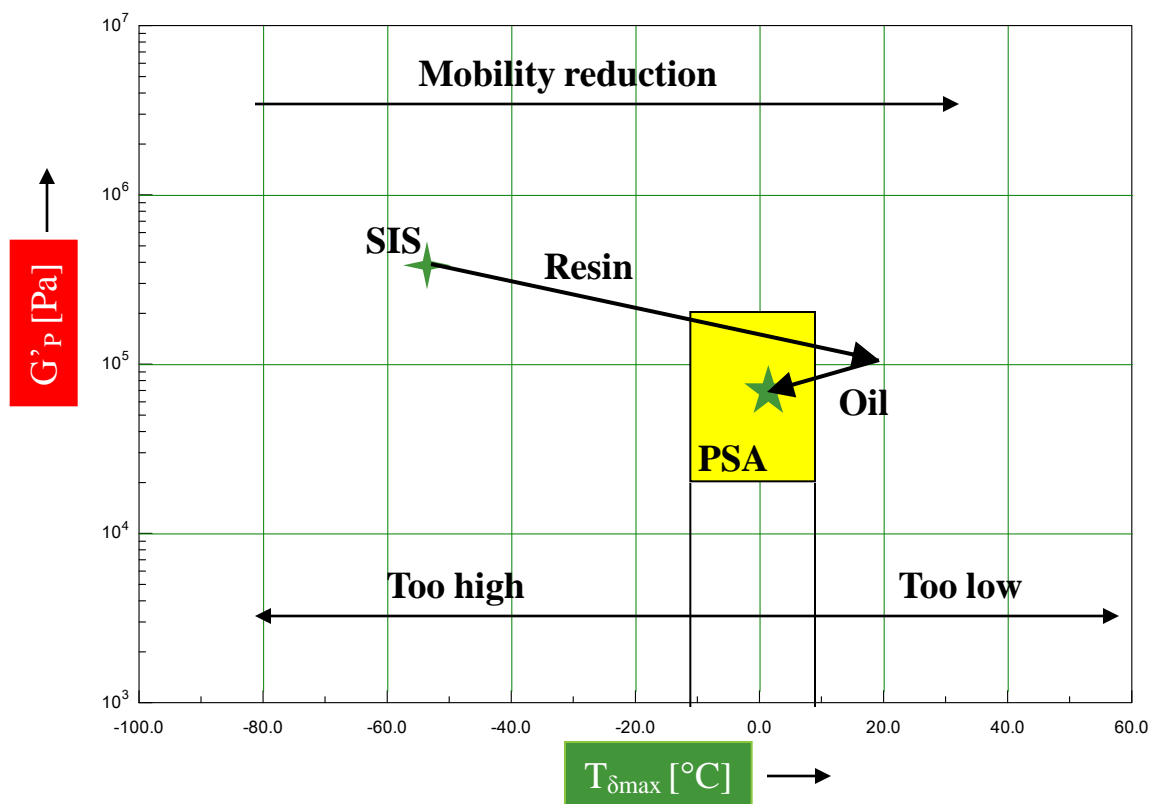


# Ideal HMPSA/ Key Parameters

## Dynamic Mechanical Analysis

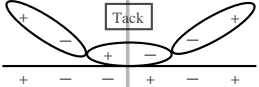
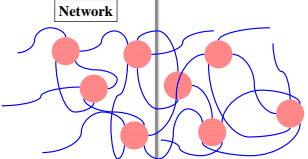
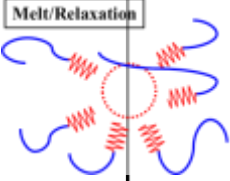
## Dahlquist !

Tack and Cohesion / Adhesion balance



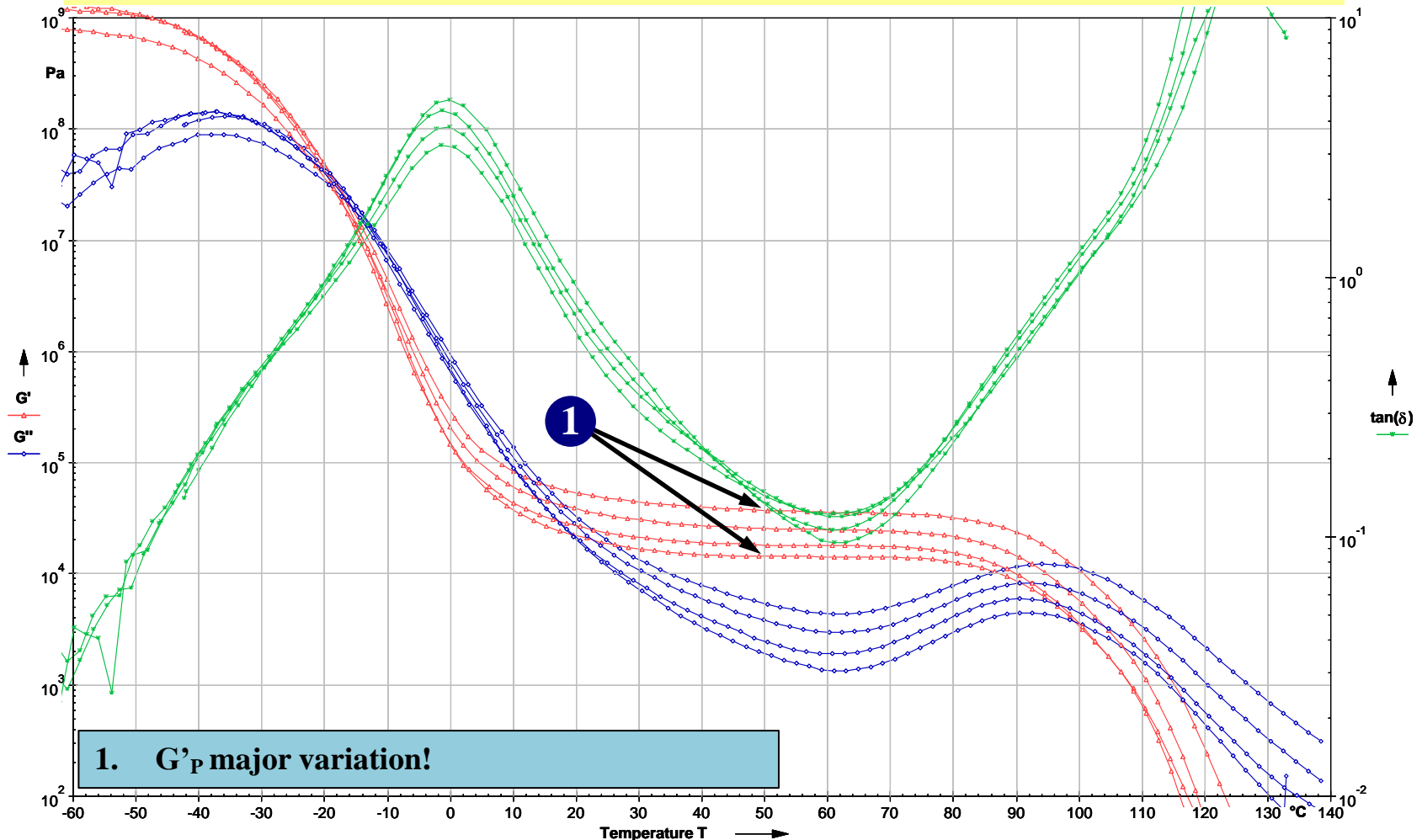
# Summary: Ideal HMPSA/ Key Parameters

## The Luth-Burgers PSA Model

	Adhesive property	“Burgers model”	“DMA control”
<b>Glassy</b>	<b>Glassy</b>	$K_1 \cong 10^9 Pa$	<i>Const.</i>
<b>Mobility/Fast processes</b>	<b>Tack/ Recovery</b>	$\eta_2$ 	$T_{\delta max}^{(*)}$
<b>Strength/ Hardness</b>	<b>Cohesion/Adhesion balance</b>	$K_2$ 	$G'_P^{(*)}$
<b>Melt Viscosity</b>	<b>Creep/ Relaxation/ Melting temp.</b>	$\eta_1$ 	$T_{cross}$

# Example A: varying (just) $G'_P$

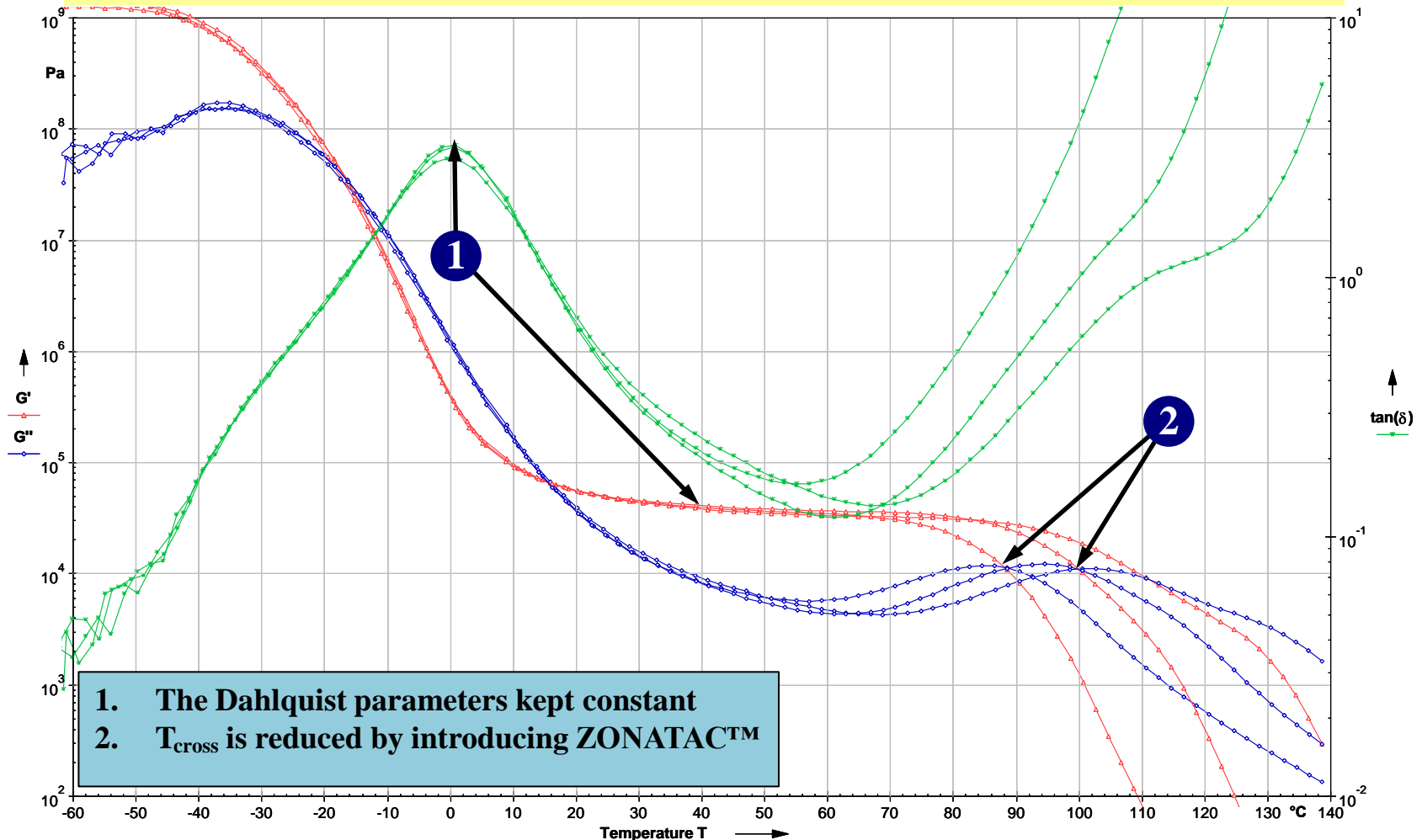
SIS/SYLVAGUM™ TR 90 /ZONATAC™ NG98 /Oil



1.  $G'_P$  major variation!

# Example B: varying (just) $T_{\text{cross}}$

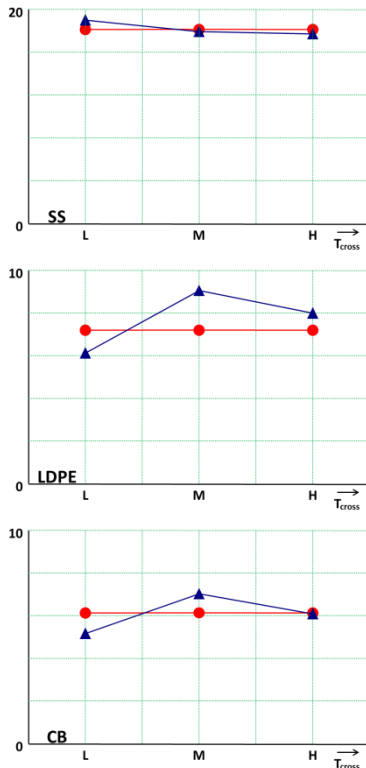
SIS/SYLVAGUM™ TR 90 /ZONATAC™ NG98 /Oil



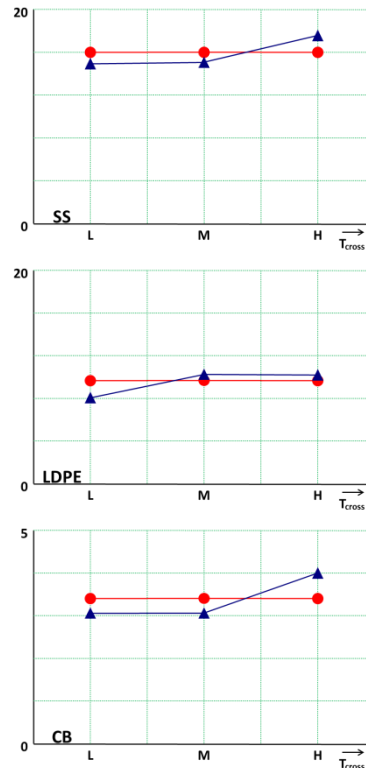
# Relating model expectations to actual values

“Outliers” well identified - deeper study

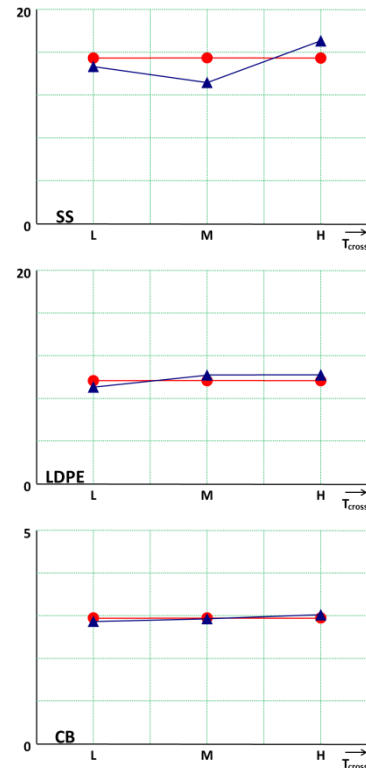
Looptack



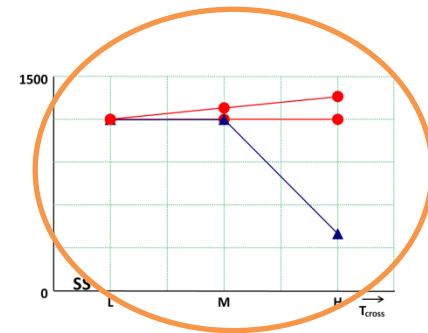
Peel 180° / 20 min



Peel 180° / 24hr



Shear @ 23°C



**Expected trend from Model**  
**Actual Data**

**Tcross controls slow processes;**  
**Loop tack and Peel constant**  
**Shear: no decrease**

# Taking to the next level



**Why measuring everything on all samples was outdated in 1920.**

*“Early in the 20<sup>th</sup> century it was proven that changing one factor at a time does not necessarily provide information about the optimum conditions”<sup>(1)</sup>.*

Let's take a smarter path, limiting the amount of experiments and using smart tools to get the most valuable data.

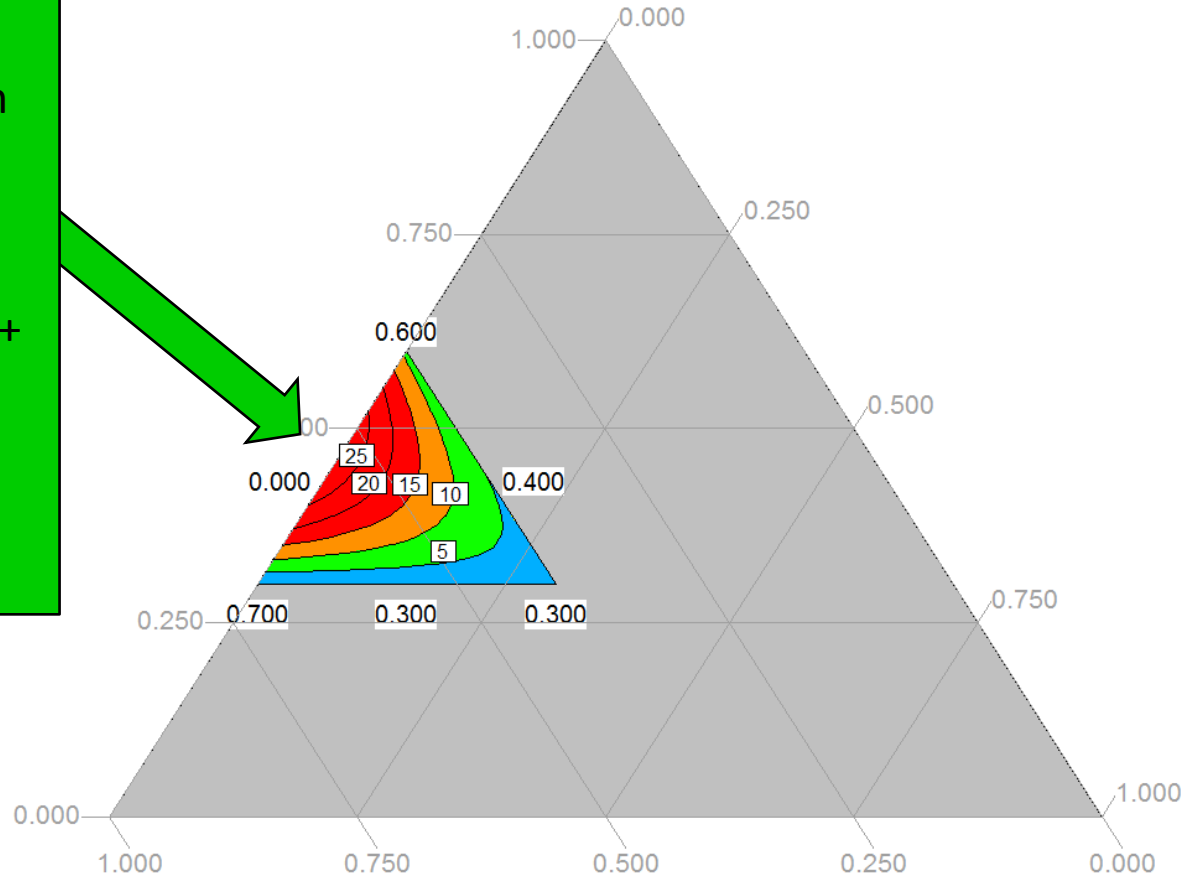
(1). Design of Experiments Principles and application – Page 3 – UMETRICS ACADEMY - 2008

# Formulation influence: adhesive properties

## Focus area

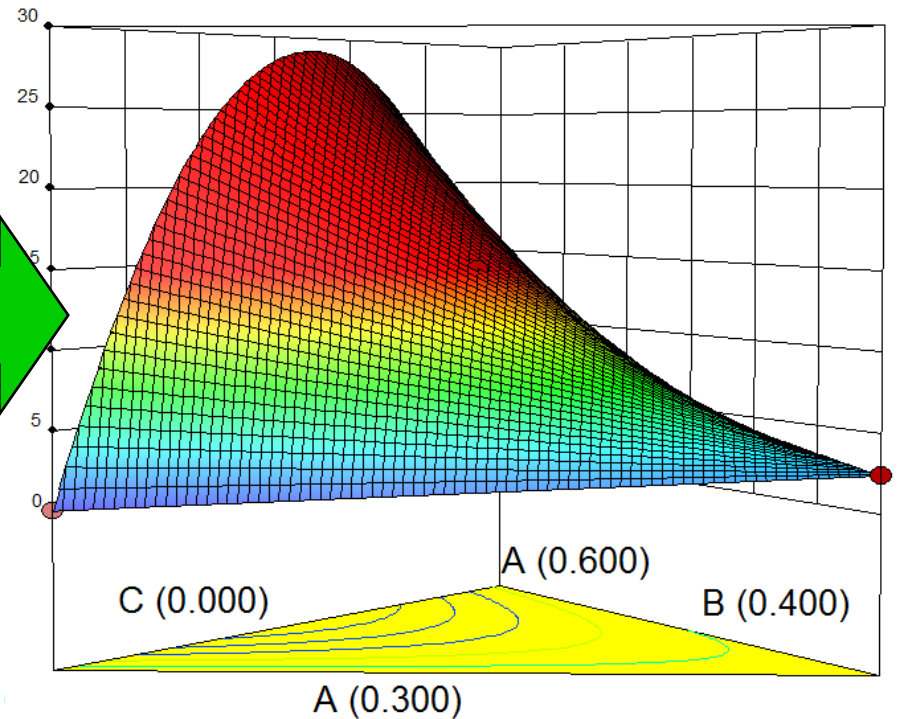
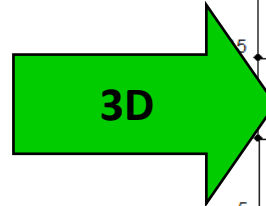
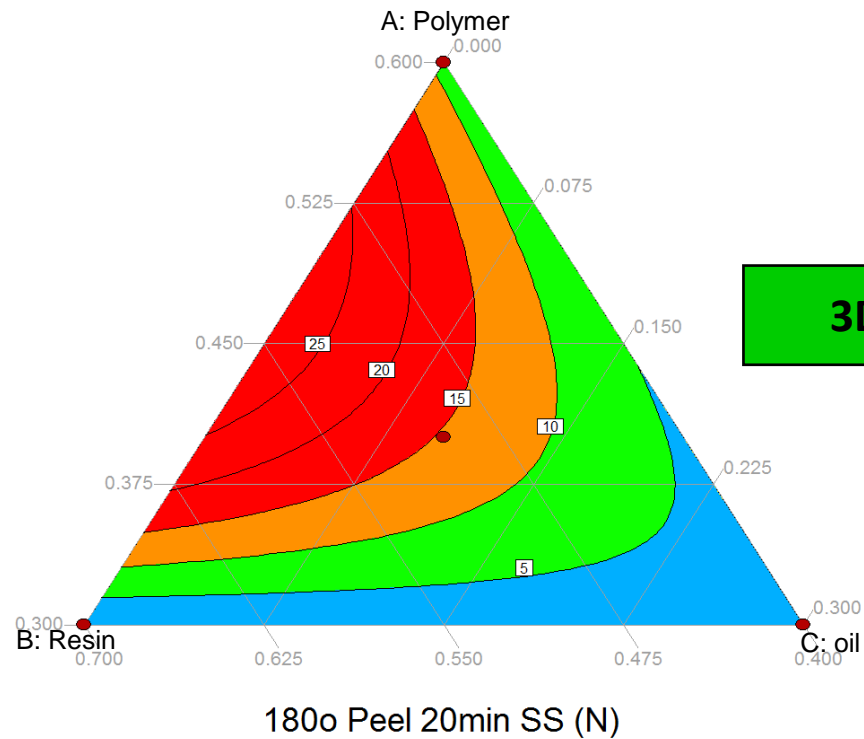
Real application data on  
formulations

- Loop Tack SS
- 180° Peel SS (20min + 24hr)
- ....



180o Peel 20min SS (N)

# Formulation influence on Peel



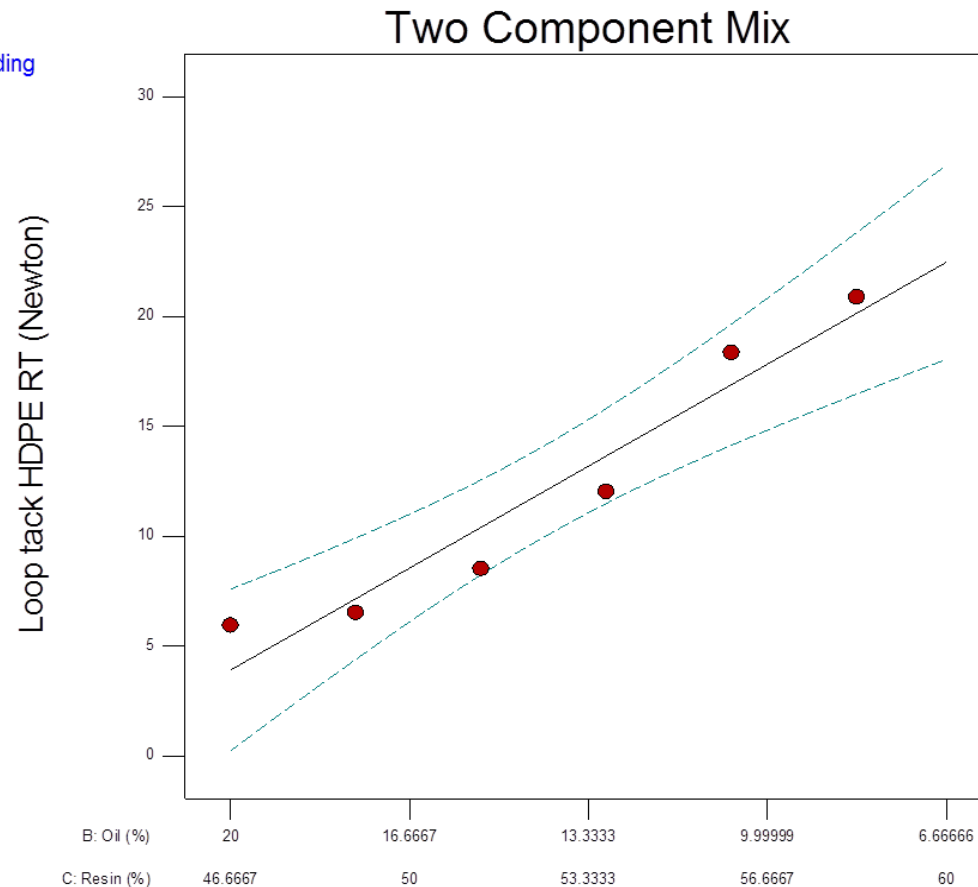


# Expanding to a DoE Model: prediction vs actual

Design-Expert® Software  
Component Coding: Actual  
Highs/Lows inverted by U\_Pseudo coding  
Loop tack HDPE RT (Newton)  
● Design Points  
— 95% CI Bands

X1 = B: Oil  
X2 = C: Resin

Purpose is to predict  
the effect of the  
resin on the  
(required) properties



# Conclusions

## Modeling of Pressure Sensitive Adhesives



- **(SBC-HM) PSA systems and the *Burgers* model have matching viscoelastic profiles**
  - Model defines the PSA key parameters.
- **The *Luth-Burgers* DMA parameters:**
  - Also describe **real** PSA systems
  - Allow a straightforward connection between viscoelastic and adhesive properties
- **LB-Model is sound basis for step 2**
  - (non-linear) interactions study using *DoE*
  - Predicting performance
- **Hypothesis: “adhesion is a bulk phenomenon” is *not* disproven**

# Contact details



- **Application Development**

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**Danny Beekman** ([danny.beekman@azchem.com](mailto:danny.beekman@azchem.com))

- **Science & Technology / Rheology**

**Roelof Luth** ([roelof.luth@azchem.com](mailto:roelof.luth@azchem.com))



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